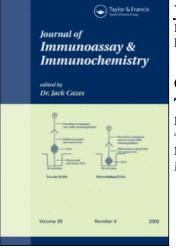
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#### COATING OF POLYMERIC SURFACES FOR IMMUNOASSAY BY A FORCED ADSORPTION TECHNIQUE

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#### ABSTRACT

A forced adsorption technique for coating polymeric surfaces is described. This technique is simple and allows us to use (1) relatively impure ligands to be coated on latex surface and (ii) the minimum amount of ligands for preparation of large-volume lots of latex reagent. The serologic results are highly reproducible from lot to lot of the reagent, in contrast to the variability found with passively adsorbed reagents. Use of this technique could be extended to coat flat surfaces for solid-phase immunoassays. (KEY WORDS: forced adsorption, latex, Rocky Mountain spotted fever)

#### INTRODUCTION

We have previously (1) reported the development of a sensitive, specific latex-<u>Rickettsia</u> <u>rickettsii</u> test for detection of antibodies to Rocky Mountain spotted fever (RMSF). The reagent for that test was prepared by passive adsorption onto latex particles (1) of erythrocyte-sensitizing substance (ESS) prepared from <u>R</u>. <u>rickettsii</u> purified by sucrose density-gradient centrifugation (2).

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To undertake an extensive interlaboratory evaluation of this test, we prepared several lots of the latex-<u>R</u>. <u>rickettsii</u> reagent using several batches of ESS. Quality control of these lots indicated a lack of reproducibility from lot to lot and a decrease in the sensitivity of the test. This prompted our concern that the diagnostic value of the test might be impaired.

To overcome this problem, we developed a forced adsorption technique which has enabled us to prepare the latex-<u>R</u>. <u>rickettsii</u> reagent in large volumes and with consistent sensitivity and reproducibility from batch to batch of ESS antigen. This note details the forced adsorption method for coating ESS onto latex particles for use in immunodiagnosis of RMSF. The method may also be applicable to the coating of other ligands on particulates or flat surfaces.

#### MATERIALS AND METHODS

#### Preparation of ESS

Ten batches of <u>R</u>. <u>ricketteii</u> ESS were prepared, nine as described by Anacker (2) and the tenth, A255, according to the method of Chang (3).

#### Forced-Adsorption Technique

Latex polystyrene particles (0.81  $\mu$ m diameter; Difco Laboratories, Detroit, MI) were adjusted with distilled water (DW) to contain 3.7 x 10<sup>10</sup> particles per ml (4). One volume of the latex suspension was centrifuged at 16,000 x g for 30 min at 5°C.

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The supernatant (labeled A), which included the manufacturer's stabilizers, was decanted, saved, and kept at 5°C for futher use to resuspend the latex pellet (4). The latex sediment was resuspended in 1/20th the volume of glycine-buffered saline (GBS; 4) pH 8.2, and stirred for 15 min with a magnetic bar until all clumps were broken.

The appropriate volume of antigen, determined by checkerboard titration, was added dropwise to the continually stirring latex mixture. The optimum volumes ranged from 15 to 50  $\mu$ l/ml. After stirring for an additional 15 min at 25°C, 8 to 10 volumes of ethanol-acetate (95% ethanol containing 0.5% sodium acetate) was added. The mixture was stirred for another 30 min at 25°C and left overnight at 5°C. The latex suspension was then centrifuged at 16,000 x g for 30 min at 5°C, and the ethanolic supernatant was discarded.

The latex sediment was resuspended in 1/10th the volume of supernatant A, and the mixture was stirred at  $25^{\circ}$ C until all clumps were broken. Supplementary ESS ( $\frac{1}{4}$  to  $\frac{1}{2}$  the volume of ESS used in coating the particles) was gradually added, and the mixture was continually stirred for an additional 30 min. The remainder of supernatant A was then added, followed by GBS, so that the combined volume of GBS and supplemental ESS was equal to the original volume of the latex suspension. Then, for every 1 volume of this suspension, 0.125 volume of GBS containing 0.1% fatty acid-free bovine albumin (GBS-BAF; Sigma Chemical Co., St. Louis, MO) was added.

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The latex-<u>R</u>. <u>rickettsii</u> suspension was dispensed in 2-ml aliquots. Each aliquot was sonicated for 15 s at a setting of 35, using the microprobe of a Fisher Sonic Dismembrator Model 300.

#### Test Sera

Individual or pooled human sera which were authenticated to contain anti-<u>R</u>. <u>rickettsii</u> antibodies by the microimmunofluorescence test of Philip et al. (5) were used to test the forcedadsorption technique. Each test serum or pool was first diluted at 1:16 in GBS-BAF and inactivated for 30 min at  $56^{\circ}$ C. The titration was performed exactly as previously described (1).

#### RESULTS

ESS antigen batches were used to prepare the latex-<u>R</u>. <u>rickettsii</u> reagent by both the passive and forced adsorption techniques. The variation in titer of a standard reactive serum assayed with reagents prepared by both techniques is shown in Table 1. Results with the passively adsorbed reagent were inconsistent. With the forced-adsorption reagents the standard serum was consistently reactive and at higher titers.

Eleven large-volume lots and seven pilot lots of latex-<u>R</u>. <u>rickettsii</u> reagent were prepared from 10 batches of ESS antigen by the forced adsorption technique and were titrated against four standard sera over a period of 3 years. The results are summarized in Table 2. For each standard the endpoints were within a single dilution with all latex-<u>R</u>. <u>rickettsii</u> lots tested, regardless of volume. Similarly all results with a given antigen

#### TABLE 1

Dependence of latex-<u>R</u>. <u>rickettsii</u> test reactivity on choice of passive or forced adsorption of ESS on latex particles<sup>a</sup>

| ESS<br>batch<br>no. | Serum titer           |                      |  |  |  |
|---------------------|-----------------------|----------------------|--|--|--|
|                     | Passive<br>adsorption | Forced<br>adsorption |  |  |  |
| A255                | 256                   | 512                  |  |  |  |
| ZR195 - 1           | 32                    | 512                  |  |  |  |
| - 2                 | NR <sup>b</sup>       | 1024                 |  |  |  |
| ZR197/198           | NR                    | 1024                 |  |  |  |

<sup>a</sup> A standard serum with a microimmunofluorescence titer of 2048 was used to test these preparations.

<sup>b</sup> NR, nonreactive.

batch were within one twofold titer difference. These findings show the reproducibility of results with the forced-adsorption reagents and the applicability of this technique for preparing large volumes (up to 680 ml) of latex-R. <u>rickettsii</u> reagent. (The 0.6-ml lots were pilot lots for the checkerboard titration. They were used to establish the optimum antigen concentration for each ESS batch, which was then used to prepare the large-volume lots).

#### DISCUSSION

Man-made polymers such as latex have been used as carriers in immunodiagnosis to detect antigens (4) or antibodies (6). The

## TABLE 2

Reproducibility of latex-R. <u>rickettsii</u> test with various lots of forced-adsorption antigen and four standard sera over a 3-year period.

| ESS<br>batch | Latex- <u>R</u> .<br>rickettsii | Volume | Tite | r of | standard<br>control | serum |
|--------------|---------------------------------|--------|------|------|---------------------|-------|
| no.          | lot no.                         | (ml)   | A    | В    | C                   | D     |
| A255         | 011179                          | 144    | 512  |      |                     |       |
| ZR195        | 011279                          | 59     | 1024 |      |                     |       |
| ZR197/198    | 011679                          | 72     | 1024 |      |                     |       |
|              | 061779                          | 0.6    | 1024 |      |                     |       |
|              | 062979                          | 325    | 1024 | 256  |                     |       |
| ZR214/215    | 012180                          | 0.6    |      | 512  |                     |       |
|              | 012280                          | 680    |      | 256  |                     |       |
| ZR220/221    | 042180                          | 558    |      |      | 64                  |       |
| ZR225/226    | 071880                          | 0.6    |      |      | 64                  |       |
|              | 072180                          | 537    |      |      | 64                  | 128   |
| ZR205        | 060181                          | 0.6    |      |      |                     | 64    |
|              | 060281                          | 269    |      |      |                     | 64    |
| ZR228        | 083181                          | 0.6    |      |      |                     | 128   |
|              | 090181                          | 124    |      |      |                     | 64    |
| ZR238/228    | 022582                          | 0.6    |      |      |                     | 128   |
|              | 030282                          | 234    |      |      |                     | 128   |
| ZR239        | 050582                          | 0.6    |      |      |                     | 128   |
|              | 051782                          | 306    |      |      |                     | 64    |

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common methods to attach these ligands to the carriers are passive adsorption, which is based on hydrophobic bonding (7), and coupling, which is based on covalent bonding (8). Both procedures require highly purified components because contaminants will compete for the limited surface area on the particle or for the active sites on activated particles. In the coupling procedure, excessive amounts of ligands are usually required to complete the reaction. This may be wasteful of ligands which are difficult to prepare. Also, the coupling procedure requires involved protocols (9) and long preparation times, which may compromise the usefulness of these reagents in test systems.

Passive adsorption is a function of the affinity (7) and relative initial concentrations (10) of the components in solution. High concentrations of impurities with high affinities for the particle surface might compete too successfully for sites on the particle and result in decreased test sensitivity. With forced adsorption the main factor is the efficiency of the precipitating agent, rather than the affinity of the components for the surface. This efficiency can be seen in the results for the ESS system (Table 1).

Erythrocyte-sensitizing substance is a product of sodium hydroxide treatment of rickettsiae without further purification. It appears to be a protein-carbohydrate complex with a proteinto-carbohydrate ratio of 2:1 (11). The carbohydrate moiety of ESS is responsible for serologic activity (11). We theorize that in passive adsorption the protein competes with the polysaccharide for the surface of the latex particles, with a resulting loss in sensitivity. In forced adsorption, however, both components are precipitated without loss of sensitivity (Table 1). Futhermore the forced adsorption method is simple, and the time required to prepare the reagent is only that needed to precipitate the ligands onto the surface.

Selective use of precipitating agents is important and is governed by several factors. The precipitating agent should precipitate the desired ligand without destroying its biologic activity or the polymeric surface to be coated. Moreover the precipitating agent should not irreversibly aggregate the particles. We have found that trichloroacetic acid, for example, cannot be used to precipitate globulin on latex because the latex pellet cannot be resuspended without drastic measures. For particulate surfaces low aliphatic alcohols, alone or mixed with low concentrations of salt (e.g. ethanol and sodium acetate), can be used successfully to precipitate proteins and carbohydrates. For ESS ethanol meets all of the essential criteria.

The quantity of antigen that can be force-adsorbed is limited by the saturation point of the particle surface. In general, once the surface is coated with ligand, there is no point in trying to force on more. Layering ligand over ligand on the particles by increasing its concentration will not optimize the sensitivity of the reagent (6, 10) because antigenic determinants in the inner layers may not be available to react. However, since the concentration of antigen adsorbed on the surface of the particle is not

ATEX)~ ESS-Ab-ESS-Ab-ESS... Ab-ESS~(LATEX

FIGURE 1. Simplified model of latex-R. <u>rickettsii</u> aggregation in the presence of supplementary souble ESS. *construction* sents ESS bound to latex particles.

usually optimum for a maximum antigen-antibody reaction (6), we add supplementary soluble ligand at a later step in the preparation of the reagent to maximize the sensitivity (6). This free ligand, e.g. <u>R. rickettsii</u> ESS, forms crosslinks between the newly formed latex-antigen/antibody complexes and free antibody to produce larger aggregates (Fig. 1), which result in higher sensitivity.

The sensitivity of the latex-<u>R</u>. <u>rickettsii</u> was much improved by the forced adsorption technique. Specificity was not adversely affected. During the development stages, when the reagent was prepared by direct adsorption, the specificity was 93.9% for single sera and approached 100% for paired sera (1). During the evaluation, when the reagent was prepared by forced adsorption, the overall specificity was 98.93% (12). The reagent prepared by forced adsorption was stable over a 17-month period with no appreciable loss of reactivity (12).

This forced adsorption technique might also be applied to prepare flat polymeric surfaces, for solid-phase immunoassay such as radioimmunoassay and enzyme immunoassay (EIA), so as to eliminate variation in test results. Kricks et al. (13) have shown substantial variation (5.2 to 29.5%) in the amount of protein adsorbed on microtiter plates for EIA by the passive adsorption method. They concluded that the variability in the amount of protein adsorbed onto the surfaces of individual wells on the same plate seriously affects the reliability and interpretation of EIA results.

In conclusion, we are presenting an alternative to the passive adsorption and coupling methods, which coat ligand onto particles in a random fashion. The forced adsorption technique enables the preparation of reproducible, stable, sensitive, and specific reagents for diagnostic test systems.

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